

**MARK SCHEME for the October/November 2008 question paper**

**0620 CHEMISTRY**

**0620/05**

Paper 5 (Practical Test), maximum raw mark 40

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- 1 Experiment 5 observations
- rapid/vigorous o.w.t.t.e (1) bubbles/fizz (1)  
blue precipitate (1) [max 2]
- Table of results  
initial boxes correctly completed (1)  
final boxes correctly completed (1)  
differences correctly completed (1) signs correct (1)  
comparable to Supervisor's results (1) [5]
- (a) all 4 bars correctly drawn (3), -1 for each incorrect labelled (1) [4]
- (b) (i) solid **A**/Experiment 1 (1) [1]  
(ii) temperature increased/heat/energy given out (1) [1]
- (c) Experiment 3/solid **C** or from results (1) [1]
- (d) (i) double the value or  $2 \times$  value from table (1) e.c.f. applies [1]  
(ii) half the value or half the value from table (1) [1]  
(iii) more/larger volume of water (1) twice as much (for solid to dissolve in) o.w.t.t.e. (1) [2]
- (e) acid present (1) carbonate present (1) carbon dioxide (1) [max 2]

**[Total: 20]**

- 2 (a) solution **K** blue/green  
solution **L** yellow/orange/brown (1) [1]
- (b) check pHs from Supervisor's results  
pH of solution **K** 1-6  
pH of solution **L** 1-6 (1) [1]
- tests on solution **K**
- (c) (i) blue (1) precipitate (1) [2]  
(ii) deep/royal blue (1) solution/precipitate dissolves (1) [2]  
(iii) no reaction/change/nothing (1) not lighter [1]  
(iv) white (1) precipitate (1) [2]

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tests on solution L

- (d)** (i) (rusty) brown (1) precipitate (1)
- (ii) brown precipitate (1) not red [1]
- (iii) no reaction/change/nothing (1) [1]
- (iv) white (1) precipitate (1) not lighter/paler [2]
- (e)** acidic (1) [1]
- (f)** chloride/  $Cl^-$  (1) [1]
- (g)** (i) copper (1) [1]
- (ii) iron (1) (III) (1) [2]

**[Total: 20]**

**[Total for paper: 40]**